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# ***DEVELOPMENTS IN THE SIMTECHE PROCESS CO<sub>2</sub> CAPTURE BY FORMATION OF HYDRATES***

**Presented to  
The Third Annual Conference on Carbon Capture and  
Sequestration  
Washington DC  
May 3-6, 2004**

# Presentation Outline

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- Background
- Objectives
- Accomplishments
- Experimental Program
- Engineering Analysis
- Program Status

# ***CO<sub>2</sub> Hydrate Program – Team Members***

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DOE/NETL Project Manager – José Figueroa  
Program Funding – Gary Steigel, Scott Klara

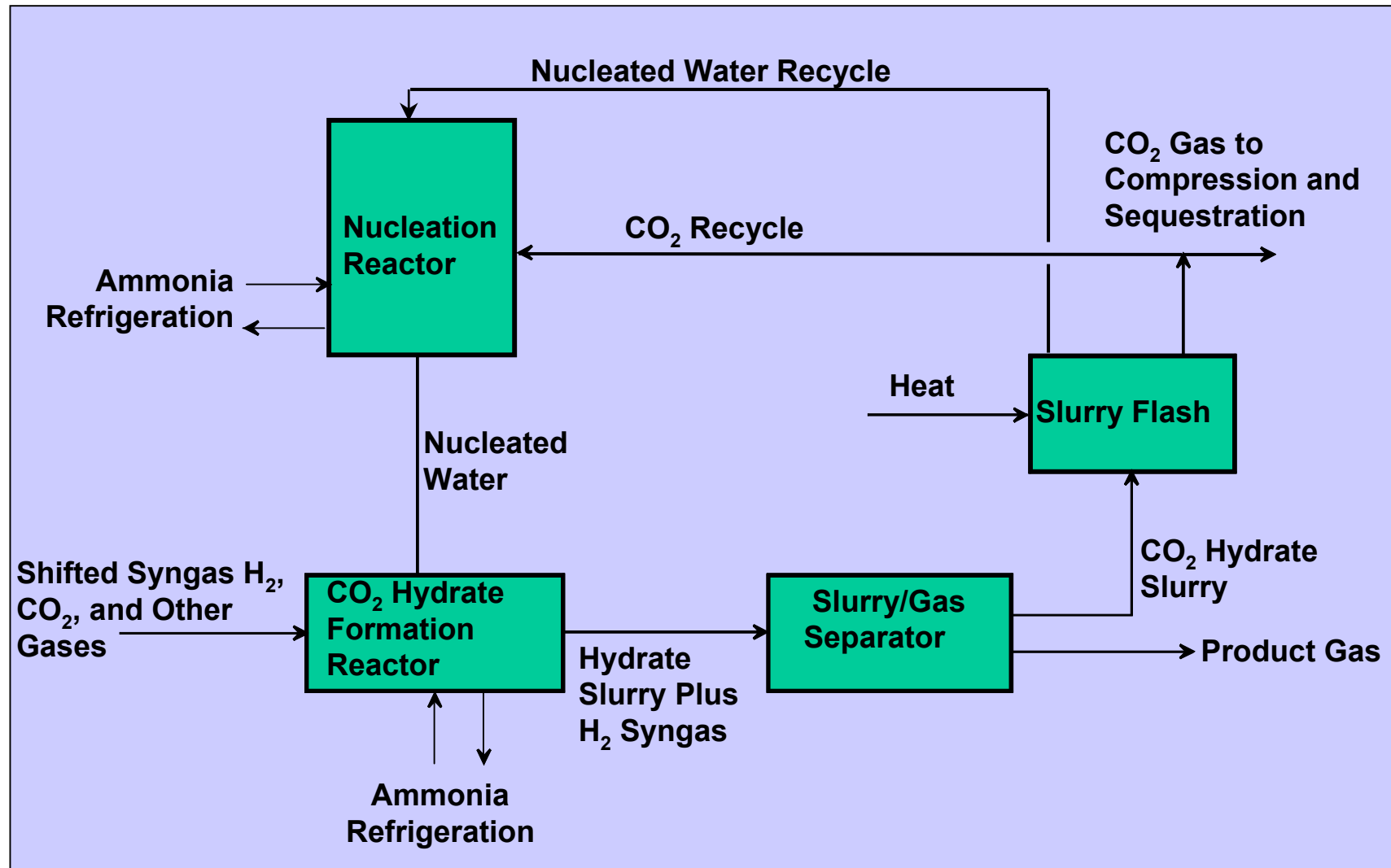
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R. Currier, G. Anderson, Dali Yang, S. Obrey  
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*Los Alamos National Laboratory*

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D. Spencer  
*SIMTECHE*

# Simteche Block Flow Diagram



## ■ Three Phases

- Phase 1 – Proof-of-Concept
- Phase 2 – Develop Scale-up Data, Design, and Build Slip-Stream Pilot Plant
- Phase 3 – Tests at Operating Gasification Plant

# Objectives

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- **Phase 2 Objectives**
  1. **Engineering Test Model**
    - Develop Design Data on Unit Operations
    - Argon/CO<sub>2</sub> Mixtures
  2. **Hydrogen Flow Tests**
    - Demonstrate Separation in Hydrogen with H<sub>2</sub>S
  3. **Design Slipstream Test Unit (STU)**
    - Skid-Mounted
    - Demonstration on Slipstream From Operating Gasifier
  
- **Phase 3 Objectives : Demonstrate Operation**

## ***Accomplishments – Phase 2***

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- **Improved Separation Prediction**
  - Correlations and calculation methods developed
- **ETM Unit Designed-Built-Operational**
  - Lessons Learned Incorporated
- **Hydrogen Flow System Relocated – Validated**
  - Initial Flow Tests Done
  - Consistency with Ar/CO<sub>2</sub> Results Established
- **STU Site Selected**
  - TECO Polk Power Plant
  - Costs Estimated
  - Design Basis Established

# ***Cost Targets***

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- **Current Estimate - \$9 per ton of CO<sub>2</sub>**
  
- **Target 1 - \$5 per ton of CO<sub>2</sub>**
  - Minimum 75% Removal
  
- **Target 2 - \$10 per ton of CO<sub>2</sub>**
  - 90+% Removal
  
- **Basis**
  - 2200 Psig Pipeline Pressure
  - High-Pressure Gasifier
  
- **Revised Economic Analysis in Progress**



# ***Revised Economic Analysis***

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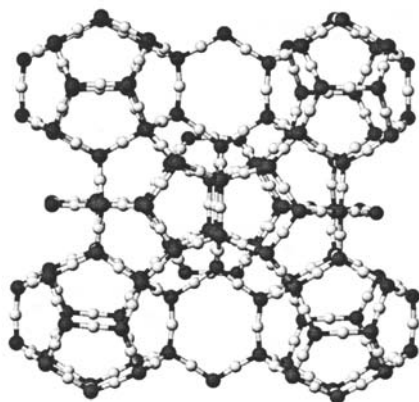
- **Incorporates NETL Analysis Standards (Developed By Scott Klara)**
  - All Technologies Compared on an Equal Basis
  - Comparison to Conventional Technologies
  - Incorporated into IGCC with Shift Reaction
  - Benchmark for All Cases: H-Class Gas Turbine
- **Design Cases**
  - Once-Thru: 70+% minimum removal
  - 2 Promoter Cases to Enhance Removal to 90%
  - Once-Thru plus Selexol to Enhance Removal to 90%

# ***Technology Development Roadmap***

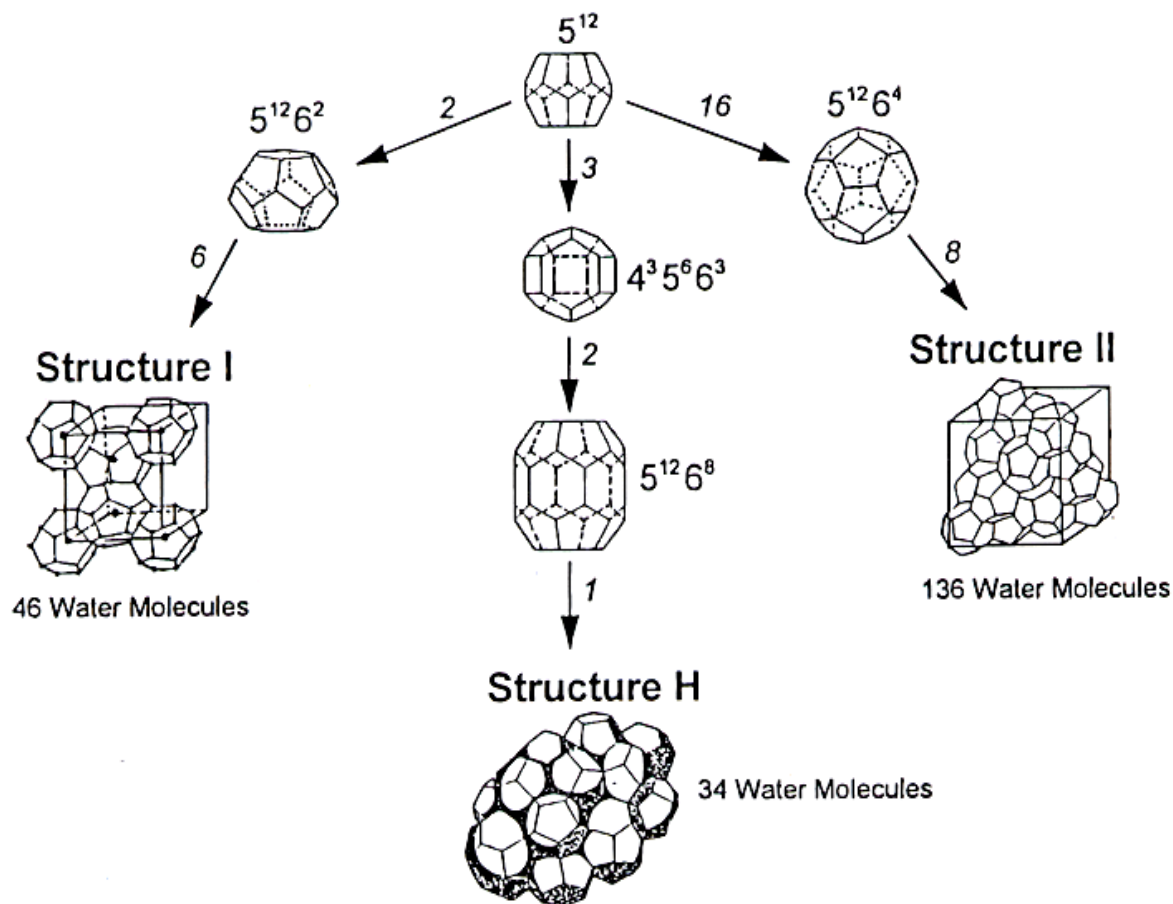
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- **Currently, configured only for high-pressure gasifiers**
  - the commercial success depends on the number of gasifiers being built
- **Promoters are being investigated for low-pressure (flue-gas) CO<sub>2</sub> capture, which broadens the market immensely**

# Ice-like Gas Hydrates



Water Structure in the sl Hydrate  
Guest Molecules (not shown) fit  
Within the polyhedral cages.



## BASIS OF SEPARATION

- CO<sub>2</sub>/H<sub>2</sub>S Form sl Hydrate
- H<sub>2</sub> Too Small to Stabilize the Water Cages

# EXPERIMENTAL ACTIVITIES

(JUNE 03- MARCH 04)



- **BENCH SCALE RIG**  
( $\text{H}_2/\text{CO}_2$ ;  $\text{H}_2/\text{CO}_2/\text{H}_2\text{S}$ )



- **E.T.M. COMPLETION**  
**AND COMMISSIONING**



- **PHASE BEHAVIOR;**  
**PROMOTER TESTING**



## BENCH SCALE FLOW TESTS ( $\text{H}_2/\text{CO}_2$ ; $\text{H}_2/\text{CO}_2/\text{H}_2\text{S}$ )

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- USE OF SURROGATE MIXES ( $\text{CO}_2/\text{Ar}$ ;  $\text{CO}_2/\text{He}$ ) VALIDATED USING  $\text{H}_2/\text{CO}_2$  MIXTURES
- SUCCESSFUL  $\text{H}_2/\text{CO}_2$  RUNS OVER THE RANGE OF 700-1000 psi
- SIMULATED SYNTHESIS GAS ( $\text{H}_2/\text{CO}_2/\text{H}_2\text{S}$ ) EXPERIMENTS

# H<sub>2</sub>/CO<sub>2</sub>/H<sub>2</sub>S FLOW TESTS

- DEPLETION OF H<sub>2</sub>S IN THE GAS RELATIVE TO CO<sub>2</sub>

**Mass Transfer Limited:** rate  $\sim k_m(C_g - C_{eq})$

Analysis of mass flux-limited hydrate growth suggests hydrate can be enriched in H<sub>2</sub>S (relative to the initial gas composition) but to an extent *less than* that measured in the equilibrium hydrate.

**Kinetically Limited:** rate  $\sim k(f_i - f_{MHF})$

Fugacity of CO<sub>2</sub> relative to its fugacity at minimum hydrate formation condition is much higher than the fugacity of H<sub>2</sub>S Relative to its minimum hydrate formation fugacity.

- EXPERIMENTAL SEQUENCE

Produce hydrate from initial gas mix. Analyze off-gas.

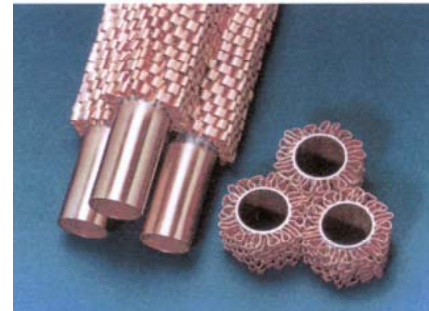
Repeat using off-gas composition and end pressure as feed.

Continue until equilibrium conversion reached.

Compare final pressure and vapor composition to equilibrium measurements/calculations.

# ENGINEERING TEST MODULE (ETM)

- REMOVE LIMITATIONS OF BENCH SCALE UNIT
  - ADDITIONAL FINNED TAIL TUBE
  - ENHANCED PROCESS COOLING SYSTEM
  - ADDITIONAL INSTRUMENTATION
  - ENHANCED WATER FEED SYSTEM



- PROVIDE ADDITIONAL DATA FOR PILOT PLANT DESIGN



# ENGINEERING TEST MODULE (ETM)





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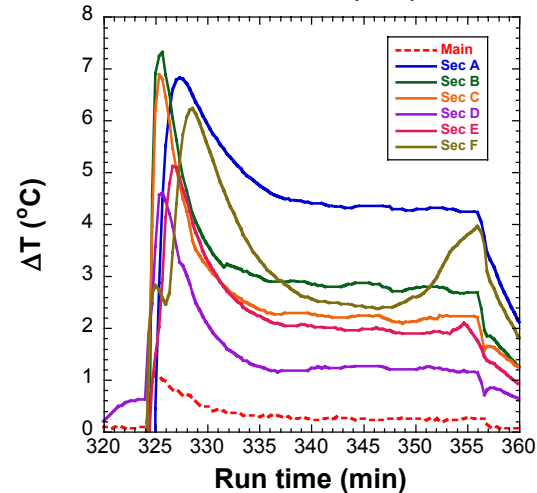
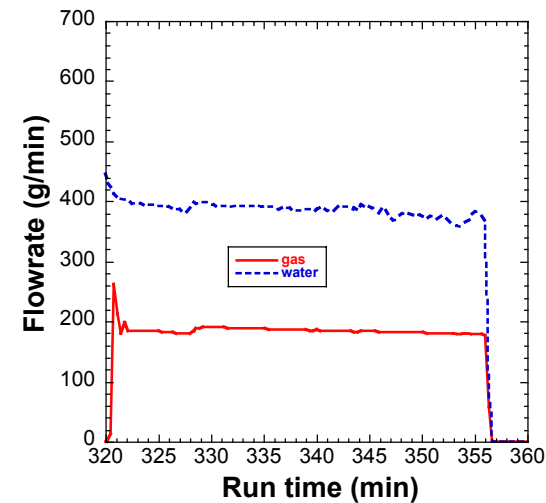
# ENGINEERING TEST MODULE (ETM)



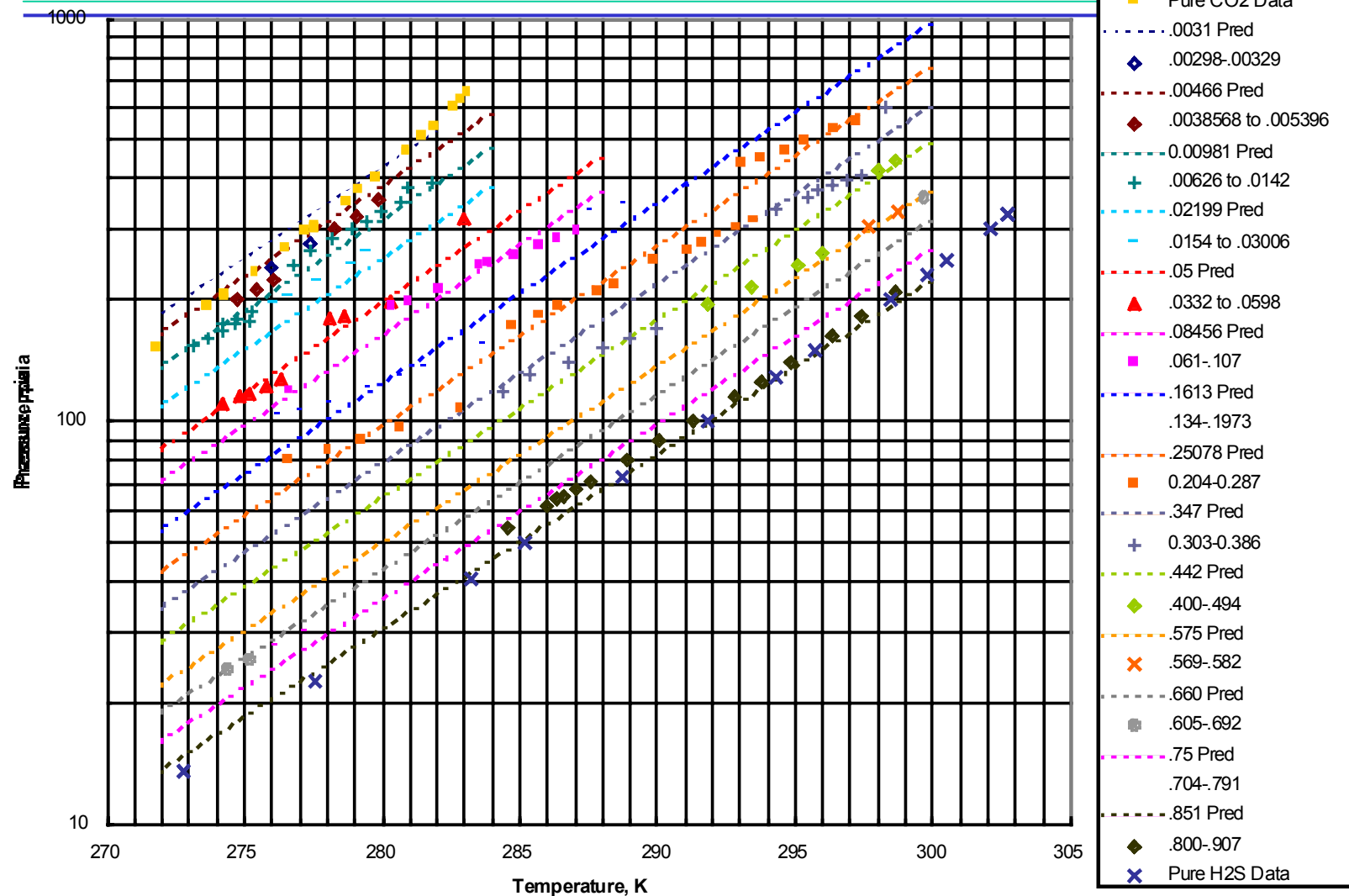
# ENGINEERING TEST MODULE (ETM)

## SYSTEM SHAKE-OUT:

- START-UP PROCEDURE DEVELOPED
- SEVERAL RUNS 30 MIN. OR MORE
- RUNS AT HIGH GAS/LIQUID RATIOS
- QUANTITATIVE RESULTS IMMINENT



# PHASE EQUILIBRIUM

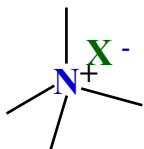




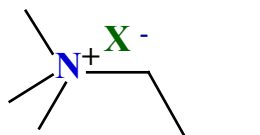
# PROMOTER DESIGN AND TESTING



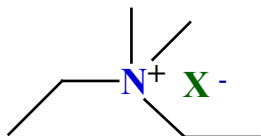
**IDEA: Employ Non-volatile Salt  
Analog of common sH-formers**



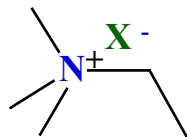
"Isopentane"



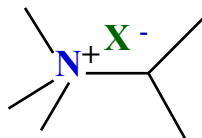
"2,2-Dimethyl pentane"



"3,3-Dimethyl pentane"

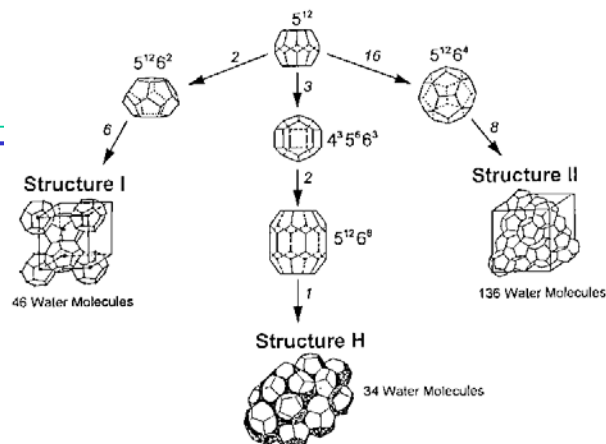


"2,2-dimethyl butane"



"2,2,3-Trimethyl butane"

*Salt analogs synthesized and  
undergoing testing with key  
constituents: CO<sub>2</sub> and CO<sub>2</sub>/H<sub>2</sub>S*



**Example of "Promoter" effect:**

(Mehta, A. P., Ph. D. Thesis, Colorado School of Mines, 1996)

**Methane requires 370 psi to form sI  
clathrate at 273K.**

**Methane-containing sH clathrate forms  
at just 140 psi with 2,2,3-Trimethyl  
butane (273K).**